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and oxygen atoms in a water molecule are polar covalent bonds. The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. The bonds between adjacent water molecules are called hydrogen bonds.

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Water Anf Aqueous Systems Guided CHAPTER 15,Water and Aqueous Systems(continued) 6.

Circle the letter next to each sentence that describes a result of the surface tension of water.

a. In a full glass of water, the water surface seems to bulge over the rim of the glass. b. Water beads up into small, nearly spherical drops on a paper towel.

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15, Water and Aqueous Systems (continued) 6. Circle the letter next to each sentence that describes a result of the surface tension of water. a. In a full glass of water, the water surface seems to bulge over the rim of the glass.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449)

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