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solution may be classified as solid, liquid or a gaseous solution. 3.

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Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute. If both components are in same physical state.

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CBSE Class 12 Chemistry Quick Revision Notes Chapter 2 Solutions. The difference in boiling points of solution and pure solvent is called elevation in boiling point . Solutions: Solutions are the homogeneous mixtures of two or more than two components. Binary solution: A solution having two components is called a binary solution.

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