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Characteristics
of Solid State

1.2 Amorphous
and Crystalline
Solids 1.3

Classification
of Crystalline
Solids 1.4

Crystal Lattices
and Unit Cells

1.5 Number of
Atoms in a Unit
Cell 1.6 Close

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Solid states
deal with

arrangements of
particles in

solids that
result in

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Several types of structure in solids. These differences in the structural unit also result in different properties of each Solid-State.

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are the chemical
substances which
are

characterised by
define shape and
volume,

rigidity, high
density, low
compressibility.

The constituent
particles

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(atoms,
molecules or
ions)

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explains the unit cell in detail, and then builds up to explain the solid state. The solutions are made in accordance with the ...

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sufficiently low
temperatures the
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to each other
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around its mean
positions and
the substance
remains in solid-
state.

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Solution.

Question: 1.1 -

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Why are solids rigid? Answer:
The particles of solids are close packed and can only oscillate about their fixed positions. These properties make solids rigid. Question:
1.2 - Why do solids have a definite volume?

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all the
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with detailed
explanation that
aims to help
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Notes . . .

Chemistry 2 In Class XI you have learnt that matter can exist in three states namely, solid, liquid and gas. Under a given set of conditions of temperature and pressure, which

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of these would be the most stable state of a given substance depends upon the net effect of two opposing factors.

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